Crystallographic report

Bis(4-nitrobenzoato)bis(pyridine)zinc(II)

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The molecular structure of $[Zn(O_2CC_6H_4NO_2-p)_2(pyridine)_2]$ exhibits a distorted N_2O_2 tetrahedral geometry around the zinc atom owing to the presence of monodentate p-nitrobenzoate ligands; the molecule has twofold symmetry. Copyright © 2004 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; zinc; *p*-nitrobenzoate; pyridine

COMMENT

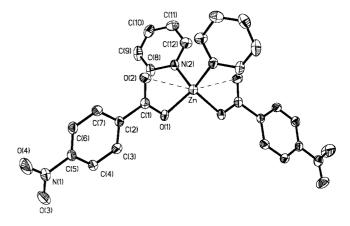
The zinc center in $[Zn(O_2CC_6H_4NO_2-p)_2(pyridine)_2]$, Fig. 1, lies on a twofold axis of symmetry and is in a distorted tetrahedral N_2O_2 coordination environment defined by two nitrogen atoms from the pyridine molecules and two carboxyl oxygen atoms derived from monodentate p-nitrobenzoate ligands. The structure is similar, for example, to those reported for $[Zn(2-pyrrolecarboxylato)_2(1-methylimidazole)_2^1$ and $[Zn(O_2CCH_3)_2(pyridine)_2^2]$

EXPERIMENTAL

An aqueous solution of ZnO (1.0 mmol) was added to a 50% ethanol solution of sodium p-nitrobenzoic acid (2.0 mmol) and pyridine (2.0 mmol) and stirred for 8.0 h at 30 °C. The white solid was obtained by filtration. The product was recrystallized from an acetonitrile solution of the compound to give colorless crystals, m.p. 166-168 °C. Intensity data were collected at 293 K on a Bruker Smart 1000 CCD for a block $0.08 \times 0.15 \times 0.25$ mm³. $C_{24}H_{18}N_4O_8Zn$, M=555.79, monoclinic, C2/c, a=14.84(3), b=6.245(12), c=24.55(5) Å, $\beta=92.14(3)$ °, V=2273(8) ų, Z=4, 1820 unique data ($\theta_{\rm max}=25.0$ °), R=0.053 (1331 data with $I>2\sigma(I)$), wR=0.168 (all data). Programs used: SHELXL and ORTEP. CCDC deposition number: 227568.

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